## Answer Key

## PHYSICS

| $10$ | b | $2$ | a | $3$ | b | $4$ | a | 5 | a | 66 | a | 7. | d | 88 | b | 9 | b | 10. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $11$ | a | $12$ | a | 13) | a | 14 | b | 15 | a | 16: | b | 17: | a | (18:) | d | 919 | c | 20 |
| $21$ | 30 | 22: | 12.00 | 23: | 3 | 24. | 18 | 25 | 2.45 |  |  |  |  |  |  |  |  |  |

## CHEMISTRY

| $10$ |  | 2 | d | 3 - | c | 4. | a | 5 | c | 6 |  | 7. |  | 88 | d | 98 | a | 10, |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $11$ | a | 12 | b | 13: | c | (14.) | c | 15: | b | (16) | c | 17: | a | 18: | b | 19: | a | 20: |
| $21$ | 4 | 22 | 3 | 23: | 21 | 24 | 10 | 25: | 5 |  |  |  |  |  |  |  |  |  |

## MATHEMATICS

| 1. | b | 2 | c | 3 | b | 4. | a | 5 | b | 68 | b | 7 | b | 8. | b | 9 | b | (10) | d |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $11$ | a | 12, | b | 13. | b | 14. | b | 15 | c | 16: | b | 17. | b | (18) | c | 19: | d | 20. | c |
| $21$ | 0.25 | 22 | 5 | 23 | 2 | 24. | 0.5 | 25 | 16 |  |  |  |  |  |  |  |  |  |  |

## Question-wise Detailed Solution

PHYSICS
2 A capillary tube of radius $r$ is lowered into a liquid of surface tension $T$ and density $\rho$. The angle of contact between the solid and the free surface of the liquid is $\theta=0^{\circ}$. During the process in which the liquid rises in the capillary, the work done by surface tension is

Solution

The liquid rise in capillary tube is :
$\mathrm{h}_{0}=\frac{2 \mathrm{~T}}{\rho \mathrm{gr}}$

The work done by surface tension will be :
$\mathrm{W}=\mathrm{Fh}_{0}=(2 \pi \mathrm{Tr})\left(\frac{2 \mathrm{~T}}{\rho \mathrm{gr}}\right)=\frac{4 \pi \pi^{2}}{\rho g}$

2 Assertion: The phase difference between any two points on a wave front is zero.

Reason: Light from the source reaches every point of the wave front at the same time.

An inverted bell, lying at the bottom of lake 47.6 m deep, has $50 \mathrm{~cm}^{3}$ of air trapped in it. The bell is brought to the surface of lake. The volume of the trapped air will become (atmospheric pressure $=70 \mathrm{~cm}$ of Hg and density of $\mathrm{Hg}=13.6 \mathrm{~g} / \mathrm{cm}^{3}$ )

Solution

According to Boyle's law, pressure and volume are inversely proportional to each other i.e. $p \propto \frac{1}{v}$

$\Rightarrow \mathrm{P}_{1} \mathrm{~V}_{1}=\mathrm{P}_{2} \mathrm{~V}_{2}$
$\Rightarrow\left(\mathrm{P}_{0}+\mathrm{h} \rho_{\mathrm{w}} \mathrm{g}\right) \mathrm{V}_{1}=\mathrm{P}_{0} \mathrm{~V}_{2}$
$\Rightarrow \quad \mathrm{V}_{2}=\left(1+\frac{\mathrm{h} \rho_{\mathrm{wg}}}{\mathrm{P}_{0}}\right) \mathrm{V}_{1}$
$V_{2}=\left(1+\frac{47.6 * 1 * 1000^{*} 10}{70^{*} 10^{-2 *} 13.6^{*} 1000 * 10}\right) V_{1}$
[As $P_{2}=P_{0}=70 \mathrm{~cm}$ of $\left.\mathrm{Hg}=70 * 10^{-2} * 13.6 * 1000 * 10\right]$
$\Rightarrow \mathrm{V}_{2}=(1+5) 50 \mathrm{~cm}^{3}=300 \mathrm{~cm}^{3}$

A radioactive sample $S_{1}$ having the activity $A_{1}$ has twice the number of nuclei as another sample $S_{2}$ of activity $A_{2}$. If $A_{2}=2 A_{1}$, then the ratio of half-life of $S_{1}$ to the half-life of $S_{2}$ is

Solution

Activity, $A=\lambda N=\frac{0.693}{T_{1 / 2}} N$

Where $T_{1 / 2}$ is the half-life of a radioactive sample,

$$
\begin{aligned}
\therefore \quad & \frac{A_{1}}{A_{2}}=\frac{N_{1}}{T_{1}} \times \frac{T_{2}}{N_{2}} \\
& \frac{T_{1}}{T_{2}}=\frac{A_{2}}{A_{1}} \times \frac{N_{1}}{N_{2}} \\
& =\frac{2 A_{1}}{A_{1}} \times \frac{2 N_{2}}{N_{2}}=\frac{4}{1}
\end{aligned}
$$

5 )
Three rods of identical cross-sectional area and made from the same metal form the sides of an isosceles triangle ABC right angled at B . The points A and B are maintained at temperatures $T$ and $\sqrt{2} T$, respectively, in the steady-state. Assuming that only heat conduction takes place, the temperature of the point $C$ is


## Solution

As $\mathrm{T}_{\mathrm{B}}>\mathrm{T}_{\mathrm{A}}$, heat flows from $B$ to $A$ through both paths $B A$ and $B C A$.
Rate of heat flow in $B C=$ Rate of heat flow in $C A$

$$
\frac{K A\left(\sqrt{2} T-T_{\mathrm{c}}\right)}{l}=\frac{K A\left(T_{\mathrm{c}}-T\right)}{\sqrt{2} l}
$$

Solving this, we get $T_{\mathrm{c}}=\frac{3 T}{\sqrt{2}+1}$


There is a thin uniform disc of radius R and mass per unit area $\sigma$, in which a hole of radius $\mathrm{R} / 2$ has been cut out as shown in the figure. Inside the hole, a square plate of same mass per unit area $\sigma$ is inserted so that its corners touch the periphery of the hole. The distance of the centre of mass of the system from the origin is


Solution

Side of square $=R \cos 45^{\circ}=\frac{R}{\sqrt{2}}$
Area of square $=\frac{\mathrm{R}^{2}}{2}$
$\mathrm{X}(\mathrm{COM})=\frac{\left(\pi \times \mathrm{R}^{2} \times \sigma \times 0+\pi \times \frac{\mathrm{R}^{2}}{4}(-\sigma) \times \frac{\mathrm{R}}{2}+\frac{\mathrm{R}^{2}}{2} \times \sigma \times \frac{\mathrm{R}}{2}\right)}{\left(\pi \times \mathrm{R}^{2} \times \sigma+\pi \times \frac{\mathrm{R}^{2}}{4}(-\sigma)+\frac{\mathrm{R}^{2}}{2} \times \sigma\right)}$
$=\frac{\mathrm{R}(2-\pi)}{2(3 \pi+2)}$
$\therefore$ The centre of mass of the system is at a distance of $\frac{\mathrm{R}(2-\pi)}{2(3 \pi+2)}$ from the centre O towards the plate as shown in the figure

7 Two cars P and Q start from a point at the same time in a straight line and their positions are represented by $X_{P}(t)=a t+b t^{2}$ and $X_{Q}(t)=f t-t^{2}$. At what time do the cars have the same velocity?

## Solution

$v_{P}=\frac{d x_{P}}{d t}=a+2 b t$
$v_{Q}=\frac{d x_{Q}}{d t}=f-2 t$
$v_{P}=v_{Q}$
$\Rightarrow a+2 b t=f-2 t$
$2 t+2 b t=f-a$
$\Rightarrow t=\frac{f-a}{2(b+1)}$

Rain falls with velocity $7 \hat{i}-10 \hat{j}$ where $x$-axis is horizontal and $y$-axis is vertical. With what velocity should a man run so that he sees rain falling vertically.

Solution
$\overline{\mathrm{v}}_{\mathrm{R} / \mathrm{m}}=\overline{\mathrm{v}}_{\mathrm{R} / \mathrm{g}}-\overline{\mathrm{v}}_{\mathrm{m} / \mathrm{g}}$
$-10 \hat{\mathrm{j}}=7 \hat{\mathrm{i}}-10 \hat{\mathrm{j}}-\overline{\mathrm{v}}_{\mathrm{m} / \mathrm{g}}$
$\overline{\mathrm{v}}_{\mathrm{m} / \mathrm{g}}=7 \hat{\mathrm{i}}$

Area of piston is $1 \mathrm{~m}^{2}$. When heat is supplied to the gas it expands and displaces piston by $\frac{\mathrm{L}}{2}$ where $\mathrm{L}=1 \mathrm{~m}$. Natural length of springs is $L=1 \mathrm{~m}$. Spring constant $K=100 \mathrm{~N} / \mathrm{m}$. The pressure of gas in final situation is - (considering equilibrium)


Solution

$\mathrm{P}=\frac{2 \mathrm{Kx}}{\mathrm{A}}$
$=\frac{2 \times 100}{1} \times \frac{1}{2}$
$=100 \mathrm{~N} / \mathrm{m}^{2}$

The primary winding of a transformer has 100 turns and its secondary winding has 200 turns. The primary is connected to an AC supply of 120 V and the current flowing in it is 10 A . The voltage and the current in the secondary are

Solution
$\frac{E_{s}}{E_{p}}=\frac{n_{s}}{n_{p}}$ or $E_{s}=E_{p} \times\left(\frac{n_{s}}{n_{p}}\right)$
$\therefore E_{s}=120 \times\left(\frac{200}{100}\right)=240 \mathrm{~V}$
$\frac{I_{p}}{I_{s}}=\frac{n_{s}}{n_{p}}$ or $I_{s}=I_{p}\left(\frac{n_{p}}{n_{s}}\right)$
$\therefore I_{s}=10\left(\frac{100}{200}\right)=5 \mathrm{~A}$

11
A ball is dropped from the top of a building. The ball takes 0.5 s to fall past the 3 m length of a window some distance from the top of the building. If the speed of the ball at the top and at the bottom of the window are $v_{T}$ and $v_{B}$ respectively, then (take $g=$ $10 \mathrm{~m} / \mathrm{s}^{2}$ ) :

Solution
$\mathrm{s}=\frac{(\mathrm{u}+\mathrm{v})}{2} \mathrm{t}$
$3=\frac{\left(\mathrm{v}_{\mathrm{T}}+\mathrm{v}_{B}\right)}{2} \times 0.5$

$\mathrm{v}_{\mathrm{T}}+\mathrm{v}_{B}=12 \mathrm{~m} / \mathrm{s}$

Also $\mathrm{v}_{\mathrm{B}}=\mathrm{v}_{\mathrm{T}}+(9.8)(0.5)$
$\mathrm{v}_{\mathrm{B}}-\mathrm{v}_{\mathrm{T}}=4.9 \mathrm{~m} / \mathrm{s}$

A deflection magnetometer is adjusted in the usual way. When a magnet is introduced, the deflection observed is $\theta$, and the period of oscillation of the needle in the magnetometer is T . When the magnet is removed, the period of oscillation is $T_{0}$. The relation between T and $T_{0}$ is

Solution

For first case
$T=2 \pi \sqrt{\frac{I}{M \sqrt{F^{2}+H^{2}}}}$

When magnet is removed
$T_{0}=2 \pi \sqrt{\frac{I}{M H}}$
Also, $\frac{F}{H}=\tan \theta$

From Eqs. (i) and (ii) we have
$\frac{T}{T_{0}}=\sqrt{\frac{H}{\sqrt{F^{2}+H^{2}}}}$
$\frac{T^{2}}{T_{0}^{2}}=\cos \theta T^{2}=T_{0}^{2} \cos \theta$

How many different wavelengths may be observed in the spectrum from a hydrogen sample if the atoms are excited to states with principal quantum number $n$ ?

## Solution

From the nth state, the atom may go to ( $n-1$ ) th state, $\ldots$. , 2 nd state of 1 st state, So there are ( $n-1$ ) possible transitions starting from the nth state. The atoms reaching $(n-1)$ th state may make $(n-2)$ different transitions. Similarly for other lower sates. The total number of possible transitions is

$$
\begin{aligned}
= & (n-1)+(n-2)+(n-3)+\ldots 2+1 \\
& =\frac{n(n-1)}{2}
\end{aligned}
$$

$(2.3+0.035+0.035) \mathrm{g}=2.37 \mathrm{~g}$

But we have to retain only one decimal place.

So, the total mass is 2.4 g

An electron having kinetic energy 10 eV is circulating in a path of radius 0.1 m in an external magnetic field of intensity $10^{-4} \mathrm{~T}$. The speed of the electron will be

Solution

When charge enters into a perpendicular magnetic field, it starts to move in a circular path.
Radius of a circular path followed by a charge particle:
$r=\frac{m V}{q B}$
$V=1.76 \times\left(10^{6}\right) \mathrm{ms}^{-1}$

A screen is placed 90 cm from an object. The image of an object on the screen is formed by a convex lens at two different locations separated by 20 cm . The focal length of the lens is

## Solution

From displacement method
$\mathrm{D}=30 \mathrm{~cm}$ and $\mathrm{d}=20 \mathrm{~cm}$
$\mathrm{f}=\frac{\mathrm{D}^{2}-\mathrm{d}^{2}}{4 \mathrm{D}} \Rightarrow \mathrm{f}=\frac{90^{2}-20^{2}}{4 \times 90}$
$\mathrm{f}=\frac{110 \times 70}{4 \times 90} \Rightarrow \mathrm{f}=21.4 \mathrm{~cm}$

The magnitude of the $X$ and $Y$ components of $\overrightarrow{\mathrm{A}}$ are 7 and 6 . Also the magnitudes of $X$ and $Y$ components of $\overrightarrow{\mathrm{A}}+\overrightarrow{\mathrm{B}}$ are 11 and 9 respectively. What is the magnitude of $\vec{B}$ ?

Solution

Let $\overrightarrow{\mathrm{A}}+\overrightarrow{\mathrm{B}}=\overrightarrow{\mathrm{R}}$. Given $A_{x}=7$ and $A_{y}=6$

Also $R_{x}=11$ and $R_{y}=9$. Therefore,
$B_{x}=R_{x}-A_{x}=11-7=4$
and $B_{y}=R_{y}-A_{y}=9-6=3$
Hence, $B=\sqrt{B_{x}^{2}+B_{y}^{2}}=\sqrt{4^{2}+3^{2}}=5$

Three identical spherical shells, each of mass $m$ and radius $r$ are placed as shown in the figure. Consider an axis XX which is touching the two shells and passing through diameter of the third shell. Moment of inertia of the system consisting of these three spherical shells about $\mathrm{XX}^{\prime}$ axis is:


Solution

Total MI of the system,
$\mathrm{I}=\mathrm{I}_{1}+\mathrm{I}_{2}+\mathrm{I}_{3}$
$\mathrm{I}_{2}=\mathrm{I}_{3}=\frac{2}{3} \mathrm{mr}^{2}+\mathrm{mr}^{2}=\frac{5 \mathrm{mr}^{2}}{3}$
$\mathrm{I}_{1}=\frac{2}{3} \mathrm{mr}^{2}$
$\therefore \mathrm{I}=2 \times 5 \frac{\mathrm{mr}^{2}}{3}+\frac{2}{3} \mathrm{mr}^{2}$
$=\frac{12 \mathrm{mr}^{2}}{3}=4 \mathrm{mr}^{2}$

19 A pendulum has angular amplitude $\theta$. Tension in the string at extreme position is $\mathrm{T}_{1}$ and at bottom is $\mathrm{T}_{2}$. If $\mathrm{T}_{2}=2 \mathrm{~T}_{1}$, then $\theta$ is equal to

## Solution

$\mathrm{T}_{1}=\mathrm{mg} \cos \theta$
$\mathrm{T}_{2}-\mathrm{mg}=\frac{\mathrm{mv}^{2}}{l}=\frac{\mathrm{m}}{l}(2 \mathrm{gh})$
$=\frac{2 \mathrm{mg}}{l}(1-\cos \theta) l$
$=2 \mathrm{mg}(1-\cos \theta) l$
or $\mathrm{T}_{2}=\mathrm{mg}+2 \mathrm{mg}(1-\cos \theta)$ $\qquad$


Given $T_{2}=2 T_{1}$
or $\mathrm{mg}+2 \mathrm{mg}(1-\cos \theta)=2 \mathrm{mg} \cos \theta$
or $\cos \theta=\frac{3}{4}$
$\therefore \theta=\cos ^{-1}\left(\frac{3}{4}\right)$

A shunt of resistance $\left(\frac{1}{m}\right)^{t h}$ of the resistance of the galvanometer is used to convert it into an ammeter. The range of the ammeter becomes.

Resistance of shunt is given by : $S=\frac{\mathrm{g}}{n-1}$ where $n=\frac{\mathrm{I}}{\mathrm{i}_{\mathrm{g}}}$
$\frac{\mathrm{g}}{m}=\frac{\mathrm{g}}{n-1}$
$\therefore n=m+1$
$\frac{\mathrm{I}}{\mathrm{i}_{\mathrm{g}}}=m+1$
$\therefore \mathrm{I}=(m+1) \mathrm{i}_{\mathrm{g}}$

$\left(\mathrm{I}-\mathrm{i}_{\mathrm{g}}\right)=(n-1) \mathrm{i}_{\mathrm{g}}$

21
A bullet of mass $M$ is fired with a velocity $50 \mathrm{~m} / \mathrm{s}$ at an angle $\theta$ with the horizontal. At the highest point of its trajectory, it collides head-on with a bob of mass 3 M suspended by a massless string of length $10 / 3 \mathrm{~m}$ and gets embedded in the bob. After the collision the string moves through an angle of $120^{\circ}$. Find the angle $\theta$.
(Take $\mathrm{g}=10 \mathrm{~m} / \mathrm{s}^{2}$ )

Solution

At the highest point, velocity of bullet is $50 \cos \theta$. So, by conservation of linear momentum
$M(50 \cos \theta)=4 M \nu$
$\nu=\left(\frac{50}{4}\right) \cos \theta$


At point $\mathrm{B}, \mathrm{T}=\mathrm{o}$ but $\nu \neq 0$
Hence, $4 \mathrm{Mg} \cos 60^{\circ}=\frac{(4 \mathrm{M}) \nu^{2}}{l}$
or $\quad \nu^{2}=\frac{\mathrm{g}}{2} l=\frac{50}{3}$
(as $l=\frac{10}{3} \mathrm{~m}$ and $\mathrm{g}=10 \mathrm{~m} / \mathrm{s}^{2}$ )
Also, $\quad \nu^{2}=\mathrm{u}^{2}-2 \mathrm{gh}$

$$
\begin{aligned}
& =\mathrm{u}^{2}-2 \mathrm{~g}\left(\frac{3}{2} l\right) \\
& =\mathrm{u}^{2}-3(10)\left(\frac{10}{3}\right)
\end{aligned}
$$

Solving Eqs. (i), (ii) and (iii), we get
$\cos \theta=0.86$ or $\theta \approx 30^{\circ}$

The equations of three waves are given by $y_{1}=A_{0} \sin (k x-\omega t), y_{2}=3 \sqrt{2} A_{0} \sin (k x-\omega t+\phi)$ and $y_{3}=4 A_{0} \cos (k x-\omega t)$.
Theses waves are in the same direction and are superimposed. The phase difference between the resultant-wave and the first wave is $\frac{\pi}{4}$ and $\phi=\frac{\pi}{n} \leq \frac{\pi}{2}$, then what is the value of $n$ ?

## Solution


$\tan \left(\frac{\pi}{4}\right)=\frac{\mathrm{BC}}{\mathrm{AC}}=\frac{\mathrm{A}_{0}(4+3 \sqrt{2} \sin \phi)}{\mathrm{A}_{0}(1+3 \sqrt{2} \cos \phi)}$

$\Rightarrow \cos \phi-\sin \phi=\frac{1}{\sqrt{2}}$
Squaring both sides, $\Rightarrow \cos ^{2} \phi+\sin ^{2} \phi-2 \cos \phi \sin \phi=\frac{1}{2}$
$\Rightarrow 2 \sin \phi \cos \phi=\frac{1}{2}$
$\Rightarrow \sin 2 \phi=\frac{1}{2}$
$\Rightarrow \phi=\frac{1}{2} \sin ^{-1}\left(\frac{1}{2}\right)=\frac{\pi}{12}$

23 A particle of mass $m$ is subjected to an attractive central force of magnitude $6(1 \mathrm{Fe}, 1 \mathrm{~W}$ and 4 O$) \frac{\mathrm{k}}{\mathrm{r}^{2}}, \mathrm{k}$ being a constant. If at instant when the particle is at an extreme position in its closed orbit, at a distance 'a' from the centre of force, its speed is $\sqrt{\left(\frac{\mathrm{k}}{2 \mathrm{ma}}\right)}$, if the distance of other extreme position is $b$. The ratio of $a / b$ is

Solution
$\mathrm{F}=-\frac{\mathrm{k}}{\mathrm{r}^{2}} ; \mathrm{U}=-\int \mathrm{Fdr}=\int \frac{\mathrm{k}}{\mathrm{r}^{2}} \mathrm{dr}=-\frac{\mathrm{k}}{\mathrm{r}}$
From conservation of momentum (angular) $\mathrm{mv}_{1} \mathrm{a}=\mathrm{mv}_{2} \mathrm{~b}$
$\Rightarrow \mathrm{v}_{2}=\frac{\mathrm{a}}{\mathrm{b}} \mathrm{v}_{1}=\frac{\mathrm{a}}{\mathrm{b}} \sqrt{\frac{\mathrm{k}}{2 \mathrm{ma}}}$
Now, from law of conservation of energy, $\mathrm{K}_{1}+\mathrm{U}_{1}=\mathrm{K}_{2}+\mathrm{U}_{2}$
$\Rightarrow \frac{1}{2} \mathrm{mv}_{1}^{2}-\frac{\mathrm{k}}{\mathrm{a}}=\frac{1}{2} \mathrm{mv}_{2}^{2}-\frac{\mathrm{k}}{\mathrm{b}}$
$\Rightarrow \frac{1}{2} \mathrm{~m}\left(\frac{\mathrm{k}}{2 \mathrm{ma}}\right)-\left(\frac{\mathrm{k}}{\mathrm{a}}\right)=\frac{1}{2} \mathrm{~m}\left(\frac{\mathrm{a}}{\mathrm{b}}\right)^{2}\left(\frac{\mathrm{k}}{2 \mathrm{ma}}\right)-\frac{\mathrm{k}}{\mathrm{b}}$
$\Rightarrow \frac{-3 \mathrm{k}}{4 \mathrm{a}}=\frac{\mathrm{ak}}{4 \mathrm{~b}^{2}}-\frac{\mathrm{k}}{\mathrm{b}}$
$\Rightarrow \mathrm{b}^{2}-\frac{4 \mathrm{a}}{3} \mathrm{~b}+\frac{\mathrm{a}^{2}}{3}=0$
Solving $\mathrm{b}=\frac{\frac{4 \mathrm{a}}{3}-\sqrt{\left(\frac{4 \mathrm{a}}{3}\right)^{2}-4\left(\frac{\mathrm{a}^{2}}{3}\right)}}{2}$
$\therefore \quad \mathrm{b}=\frac{\frac{4 \mathrm{a}}{3}-\frac{2 \mathrm{a}}{3}}{2}=\frac{\mathrm{a}}{3}$
$\therefore \quad \frac{a}{b}=3$

Two capacitors A and B with capacities $3 \mu \mathrm{~F}$ and $2 \mu \mathrm{~F}$ are charged to a potential difference of 100 V and 180 V respectively. The plates of the capacitors are connected as shown in the figure with one wire of each capacitor free. The upper plate of A is
positive and that of $B$ is negative. An uncharged $2 \mu \mathrm{~F}$ capacitor C with lead wires falls on the free ends to complete the circuit. Calculate the amount of electrostatic energy stored in the system after completion of the circuit.


## Solution

Charge on capacitor A , before joining with an uncharged capacitor $q_{A}=C V=(100)(3) \mu C=300 \mu C$ Similarly, charge on capacitor $B$ $q_{B}=(180)(2) \mu \mathrm{C}=360 \mu \mathrm{C}$


Let $q_{1}, q_{2}$ and $q_{3}$ be the charges on the three capacitors after joining them as shown in figure.
( $\mathrm{q}_{1}, \mathrm{q}_{2}$ and $\mathrm{q}_{3}$ are in microcoulombs)
From conservation of charge

Net charge on plates 2 and 3 before joining
$=$ net charge after joining $30=q_{1}+q_{2}$

Similarly, net charge on plates 4 and 5 before joining
= net charge after joining
$-360=-q_{2}-q_{3}$ or $360=q_{2}+q_{3}$

Applying Kirchhoff's second law in closed loop ABCDA
$\frac{q_{1}}{3}-\frac{q_{2}}{2}+\frac{q_{3}}{2}=0$
or $2 q_{1}-3 q_{2}+3 q_{3}=0$

Solving Eqs. (i), (ii) and (iii), we get

$$
q_{1}=90 \mu \mathrm{C}, \mathrm{q}_{2}=210 \mu \mathrm{C}
$$

and

$$
q_{3}=150 \mu \mathrm{C}
$$

Electrostatic energy stored after, completing the circuit
$U_{f}=\frac{1}{2} \frac{\left(90 \times 10^{-6}\right)^{2}}{\left(3 \times 10^{-6}\right)}+\frac{1}{2} \frac{\left(210 \times 10^{-6}\right)^{2}}{\left(2 \times 10^{-6}\right)}$
$=18 \times \mathrm{lo}^{-2} \mathrm{~J}$
or $\mathrm{U}_{\mathrm{f}}=18 \mathrm{~mJ}$

An uncalibrated spring balance is found to have a period of oscillation of 0.314 s , when a 1 kg weight is suspended from it, how much does the spring elongate (in cm ), in when a 1 kg weight is suspended from it ? Take $\pi=3.14$

Here, $\mathrm{T}=0.314 \mathrm{~s} ; \mathrm{m}=1 \mathrm{~kg}$

Now, $\mathrm{T}=2 \pi \sqrt{\frac{m}{k}} \quad$ or $\quad k=\frac{4 \pi^{2} m}{\mathrm{~T}^{2}}$
or

$$
k=\frac{4 \pi^{2} \times 1}{(0.314)^{2}}=\frac{4 \times(3.14)^{2} \times 1}{(0.314)^{2}}=400 \mathrm{~N} \mathrm{~m}^{-1}
$$

When spring is loaded with a weight 1 kg ,
$\mathrm{mg}=\mathrm{k} l$
or $\quad l=\frac{\mathrm{mg}}{k}$
or

$$
l=\frac{1 \times 9.8}{400}=0.0245 \mathrm{~m}=2.45 \mathrm{~cm}
$$

CHEMISTRY

Conductometric titration curve of an equimolar mixture of a HCl and HCN with $\mathrm{NaOH}(\mathrm{aq})$ is :

## Solution

Molar conductivity of $\mathrm{H}^{+}$and $\mathrm{OH}^{-}$are very high as compare to other ions. Initially, conductance of solution sharply decreases due to consumption of free $\mathrm{H}^{+}$then increases due to formation of salt ( NaCN ) and after complete neutralization, further sharply increases due to presence of $\mathrm{OH}^{-}$.

2 Which of the following carbides produces propyne on reaction with water?

## Solution

(a) $\mathrm{CaC}_{2}+\mathrm{H}_{2} \mathrm{O} \quad \rightarrow \quad \mathrm{Ca}(\mathrm{OH})_{2}+\mathrm{HC} \equiv \mathrm{CH} \uparrow$
(b) $\mathrm{Be}_{2} \mathrm{C}+2 \mathrm{H}_{2} \mathrm{O} \rightarrow \quad \rightarrow \quad 2 \mathrm{BeO}+\mathrm{CH}_{4} \uparrow$
(c) $\mathrm{Al}_{4} \mathrm{C}_{3}+12 \mathrm{H}_{2} \mathrm{O} \rightarrow \quad 4 \mathrm{Al}(\mathrm{OH})_{3}+3 \mathrm{CH}_{4} \uparrow$
(d) $\mathrm{Mg}_{2} \mathrm{C}_{3}+4 \mathrm{H}_{2} \mathrm{O} \rightarrow 2 \mathrm{Mg}(\mathrm{OH})_{2}+\mathrm{H}_{3} \mathrm{C}-\mathrm{C} \equiv \mathrm{CH} \uparrow$

An organic compound contains $49.3 \%$ carbon, $6.84 \%$ hydrogen and its vapour density is 73 . What is the molecular formula of the compound?

Solution
$\%$ of oxygen $=100-(49.3+6.84)$
$=43.84 \%$

| Element | $\%$ | At. wt. | $\frac{\text { Percentage }}{\text { at. wt }}$ | Simplest ratio |
| :--- | :--- | :--- | :--- | :--- |
| C | 49.30 | 12 | $\frac{49.30}{12}=4.11$ | $\frac{4.11}{2.74}=1.5 \times 2=3$ |
| H | 6.84 | 1 | $\frac{6.84}{1}=6.84$ | $\frac{6.84}{2.74}=2.5 \times 2=5$ |
| O | 43.84 | 16 | $\frac{43.84}{16}=2.74$ | $\frac{2.74}{2.74}=1.0 \times 2=2$ |

Hence, empirical formula $=\mathrm{C}_{3} \mathrm{H}_{5} \mathrm{O}_{2}$

Empirical formula weight
$=12 \times 3+5 \times 1+16 \times 2$
$=73$

$$
\begin{aligned}
& =2 \times 73 \\
& =164 \\
\therefore \quad \mathrm{n} & =\frac{\text { molecular weight }}{\text { empirical formula weight }} \\
& =\frac{146}{73} \\
& =2
\end{aligned}
$$

$\therefore$ Molecular formula of compound

$$
\begin{aligned}
& =\left(\mathrm{C}_{3} \mathrm{H}_{5} \mathrm{O}_{2}\right)_{2} \\
& =\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{O}_{4}
\end{aligned}
$$

The temperature, at which a gas shows maximum ideal behaviour, is known as

## Solution

Boyles' temperature is the temperature at which a real gas exhibit ideal behaviour for considerable range of pressure. It is related with van der Waals' constant as
$T_{B}=\frac{a}{b R}$

Only Iodine forms Hepta-fluoride $\mathrm{IF}_{7}$, but Chlorine and Bromine give Penta-fluorides. The reason for this is :

## Solution

Due to larger size of iodine atom, it can accommodate upto seven small fluorine atoms around, while due to smaller sizes of chlorine and bromine atoms cannot accommodate seven fluorine atoms, i.e., steric factor dominate in case of chlorine and bromine.

The following data were obtained from the first order decomposition of $2 \mathrm{~A}(\mathrm{~g}) \rightarrow \mathrm{B}(\mathrm{g})+\mathrm{C}(\mathrm{S})$ at a constant volume and at a particular temperature

| S No. | Time | Total pressure in Pascal |
| :--- | :--- | :--- |
| 1 | At the end of 10 min | 300 |
| 2 | After completion | 200 |

The rate constant in $\min ^{-1}$ is

Solution

$$
\mathrm{t}=0
$$

| 2 A | $\rightarrow$ | B |
| :---: | :---: | :---: |
| 2 P |  | 0 |
| $2 \mathrm{C}-2 \mathrm{x}$ |  | - |
| 0 |  | - |
|  |  | - |

Given, $\mathrm{P}=200$ \{after the completion\}
$2 \mathrm{P}-\mathrm{x}=300$ (at the end of 10 min )
$\mathrm{x}=100$
$\mathrm{k}=\frac{2.303}{10} \log \frac{200}{100}=0.0693 \mathrm{~min}^{-1}$
$\mathrm{Ln}^{3+}$ compounds are generally coloured in the solid state as well as in aqueous soloution.colour appears due to presence of unpaired $f$-electrons which undergo $f$ - $f$ transition.
s-electrons of the valence shell of some elements show reluctance in bond formation. Such elements are $\qquad$ and belong to
;

## Solution

Heavy p -Block elements of group $\mathrm{III}_{\mathrm{A}}, \mathrm{IV}_{\mathrm{A}}, \mathrm{V}_{\mathrm{A}}$ and in periods No .5 \& No. 6 with configuration $\mathrm{ns}^{2} \mathrm{np}^{1-3}$. The ns electrons due to close proximity to nucleus do not participate in electrovalency, only p-electrons do so.

A new Iron containing compound can have either of the two possible formulae
$\mathrm{K}_{3}\left[\mathrm{Fe}\left(\mathrm{C}_{2} \mathrm{O}_{4}\right)_{3}\right]$ or $\mathrm{K}_{2}\left[\mathrm{Fe}\left(\mathrm{C}_{2} \mathrm{O}_{4}\right)_{2} \cdot\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\right] \cdot \mathrm{A} 1.356 \mathrm{~g}$ of compound in acid is dissolved which converts oxalate to oxalic acid the solution required 34.5 mL of $0.108 \mathrm{M}_{\mathrm{KnO}}^{4}$ to reach point of equivalence. Which is correct formula of compound. Report its molar mass

Solution

Meqs of $\mathrm{KMnO}_{4}=34.5 \times 0.108 \times 5=18.63$ Meqs of compound

Equivalent weight of compound $=\frac{1.356}{18.63} \times 1000=72.78$
Redox changes are

$$
\begin{array}{ccc}
{\left[\mathrm{Fe}\left(\mathrm{C}_{2} \mathrm{O}_{4}\right)_{3}\right]^{3-}} & \cdots \mathrm{Fe}^{3+}+6 \mathrm{CO}_{2}+6 \mathrm{e}^{-} \\
{\left[\mathrm{Fe}\left(\mathrm{C}_{2} \mathrm{O}_{4}\right)_{2}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\right]^{2-}} & \cdots & \mathrm{Fe}^{3+}+4 \mathrm{CO}_{2}+5 \mathrm{e}^{-}
\end{array}
$$

Equivalent weights are
for $\quad \mathrm{K}_{3}\left[\mathrm{Fe}\left(\mathrm{C}_{2} \mathrm{O}_{4}\right)_{3}\right] \quad=\frac{437}{6}=72.8$
for $\mathrm{K}_{2}\left[\mathrm{Fe}\left(\mathrm{C}_{2} \mathrm{O}_{4}\right)_{2}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\right]=\frac{346}{5}=69.2$
i.e., compound is $\mathrm{K}_{3}\left[\mathrm{Fe}\left(\mathrm{C}_{2} \mathrm{O}_{4}\right)_{3}\right]$ and its molar mass $=437 \mathrm{~g} /$ mole.

Which among the following statements is false ?

## Solution

(i) Osmotic pressure $\pi=\mathrm{i}$ CRT
$\therefore$ concentration is same so $\pi \propto \mathrm{i}$ (number of ions)
i for $\mathrm{BaCl}_{2}=3, \mathrm{KCl}=2, \mathrm{CH}_{3} \mathrm{COOH} \leq 1<\mathrm{i} \leq 2$ and for sucrose $=1$

So, order of osmotic pressure $\mathrm{BaCl}_{2}>\mathrm{KCl}>\mathrm{CH}_{3} \mathrm{COOH}>$ Sucrose
Acetic acid does not dissociate completely. Hence, its van't Hoff factor will be in between 1 and 2 .
(ii) The extent of depression in freezing point varies with the number of solute particles for a fixed solvent only and it's a
characteristic feature of the nature of solvent also $\Delta \mathrm{T}_{\mathrm{f}}=\mathrm{k}_{\mathrm{f}} \times \mathrm{m}$
For different solvents, value of $\mathrm{k}_{\mathrm{f}}$ is also different so, for two different solvents the extent of depression may vary even if same number of solute particles be dissolved in them.

[^0]${ }_{90} \mathrm{Th}^{228} \underset{-4 \alpha}{\longrightarrow} 82 \mathrm{Th}^{212} \underset{-\beta}{\rightarrow} \underset{\text { Daughter }}{83 \mathrm{Th}^{212}}$
element

Number of neutrons = Mass number - Atomic number
$=212-83=129$

What volume of $1.00 \mathrm{~mol} \mathrm{~L}^{-1}$ aqueous sodium hydroxide is neutralized by 200 mL of $2.00 \mathrm{~mol} \mathrm{~L}^{-1}$ aqueous hydrochloric acid? Find the mass of sodium chloride produced. The Neutralization reaction is NaOH (aq.) +HCl (aq.) $\rightarrow \mathrm{NaCl}($ aq. $)+\mathrm{H}_{2} \mathrm{O}(\mathrm{l})$.

Solution

Meq. of $\mathrm{NaOH}=$ Meq. of $\mathrm{HCl}=$ Meq. of NaCl
$\mathrm{M} \times \mathrm{V}_{\mathrm{mL}}=\mathrm{M} \times \mathrm{V}_{\mathrm{mL}}\left(\mathrm{N}_{\mathrm{HCl}}=\mathrm{M}_{\mathrm{HCl}} ; \mathrm{N}_{\mathrm{NaOH}}=\mathrm{N}_{\mathrm{NaOH}}\right)$
$1 \times V=2 \times 200$
$\therefore \quad \mathrm{V}=400 \mathrm{~mL}$

Millimoles. of $\mathrm{NaCl}=2 \times 200=400$
$\therefore \quad \frac{\mathrm{w}}{58.3} \times 1000=400$
$\therefore \quad \mathrm{w}_{\mathrm{NaCl}}=23.4 \mathrm{~g}$
(73)2-Methylpent-2-ene on reductive ozonlysis will give

Solution


Which one of the following constitutes a group of the isoelectronic species?

## Solution

Species having the same number of electrons called as isoelectronic species. By calculating the number of electrons in each species given here, we get
$\mathrm{CN}^{-}(6+7+1=14) ; \mathrm{N}_{2}(7+7=14) ;$
$\mathrm{O}_{2}^{2-}(8+8+2=18) ; \mathrm{C}_{2}^{2-}(6+6+2=14) ;$
$\mathrm{O}_{2}^{-}(8+8+1=17) ; \mathrm{NO}^{+}(7+8-1=14)$
$\mathrm{CO}(6+8=14) ; \mathrm{NO}(7+8=15)$
From the above calculation we find that all the species listed in choice $c$ have 14 electrons each so it is the correct answer.

The correct IUPAC name of the following compounds is



2-cyclopropyl-3-methylbutane.
Longest chain is the open chain not the cyclic one. Numbering is done in alphabetical order of locants since they are equidistant from the terminals.

Which one of the following is manufactured by the electrolysis of fused sodium chloride

## Solution

$\underset{\text { (fused) }}{2 \mathrm{NaCl}} \rightarrow 2 \mathrm{Na}^{+}+2 \mathrm{Cl}^{-}$

Anode: $2 \mathrm{Cl}^{-} \rightarrow 2 e^{-}+C l_{2}$ (oxidation)

Cathode: $2 \mathrm{Na}^{+}+2 e^{-} \rightarrow 2 \mathrm{Na}$ - (reduction)

The solubility product of AgCl is $1.8 \times 10^{-10}$. Precipitation of AgCl will occur only when equal volumes of which of the following solutions are mixed?

Solution

After mixing,
$\left[\mathrm{Ag}^{+}\right]=\frac{1}{2} \times 10^{-4}=5 \times 10^{-5} \mathrm{M}$
$\left[\mathrm{Cl}^{-}\right]=\frac{1}{2} \times 10^{-4}=5 \times 10^{-5} \mathrm{M}$
$\mathrm{Q}_{\mathrm{sp}}=\left[\mathrm{Ag}^{+}\right]\left[\mathrm{Cl}^{-}\right]=\left(5 \times 10^{-5}\right)^{2}=2.5 \times 10^{-9}$
Since, $Q_{\text {sp }}>K_{\text {sp }} \rightarrow$ precipition takes place

A body centred cubic lattice is made up of hollow sphere of B. Sphere of solid A are present in hollow sphere of B. Radius of A is half of the radius of $B$. What is the ratio of total volume of sphere B unoccupied by $A$ in unit cell and volume of unit cell?

## Solution

Effective number of atoms of $B$ present in a unit cell $=2$

Total volume of $B$ unoccupied by $A$ in a unit cell
$=2 \times \frac{4}{3}\left(\mathrm{R}^{3}-\mathrm{r}^{3}\right) \times \pi$
$=\frac{7}{3} \pi \mathrm{R}^{3}\left(\mathrm{r}=\frac{\mathrm{R}}{2}\right)$

Volume of unit cell $=a^{3}$
$\left(\frac{4 \mathrm{R}}{\sqrt{3}}\right)^{3}=\frac{64}{3 \sqrt{3}} \mathrm{R}^{3}(\sqrt{3} \mathrm{a}=4 \mathrm{R})$
Ratio of total volume of sphere $B$ unoccupied by $A$ in unit cell and volume of unit cell $=\frac{7 / 3 \pi \mathrm{R}^{3}}{\frac{64}{3 \sqrt{3}} \mathrm{R}^{3}}=\frac{7 \pi \sqrt{3}}{64}$
19. For an ideal gas $\frac{\mathrm{C}_{\mathrm{p}, \mathrm{m}}}{\mathrm{C}_{\mathrm{v}, \mathrm{m}}}=\gamma$. The molecular mass of the gas is M , its specific heat capacity at constant volume is :

Solution
$\because \frac{\mathrm{C}_{\mathrm{p}, \mathrm{m}}}{\mathrm{C}_{\mathrm{v}, \mathrm{m}}}=\gamma$ and $\mathrm{C}_{\mathrm{p}, \mathrm{m}}-\mathrm{C}_{\mathrm{v}, \mathrm{m}}=\mathrm{R}$
$\therefore \mathrm{C}_{\mathrm{v}, \mathrm{m}}=\frac{\mathrm{R}}{\gamma-1}$
$\mathrm{C}_{\mathrm{v}, \mathrm{m}}=\frac{\mathrm{C}_{\mathrm{v}}}{\mathrm{n}}$ and $\mathrm{C}_{\mathrm{v}}=\mathrm{m} . \mathrm{c}_{\mathrm{v}}$
$\therefore \frac{\mathrm{R}}{\gamma-1}=\frac{\mathrm{m} \cdot \mathrm{c}_{\mathrm{v}}}{\mathrm{m}} \times \mathrm{M}$ and
$\therefore \mathrm{C}_{\mathrm{v}}=\frac{\mathrm{R}}{(\gamma-1) \mathrm{M}}$

A sample of pure $\mathrm{PCl}_{5}$ was introduced into an evacuated vessel at 473 K . After equilibrium was attained, concentration of $\mathrm{PCl}_{5}$ was found to be $0.05 \mathrm{~mol} \mathrm{~L}^{-1}$. If value of Kc is $8.3 \times 10^{-3}$, what are the concentration of $\mathrm{PCl}_{3}$ and $\mathrm{Cl}_{2}$ at equilibrium?
$\mathrm{PCl}_{5}(\mathrm{~g}) \rightleftharpoons \mathrm{PCl}_{3}(\mathrm{~g})+\mathrm{Cl}_{2}(\mathrm{~g})$
Solution
$\mathrm{PCl}_{5}(\mathrm{~g}) \rightleftharpoons \mathrm{PCl}_{3}(\mathrm{~g})+\mathrm{Cl}_{2}(\mathrm{~g})$
Given, $\left[\mathrm{PCl}_{5}\right]_{\text {equili. }}=0.05 \mathrm{~mol} \mathrm{~L}^{-1}$
$\mathrm{Kc}=8.3 \times 10^{-3}$

$$
\mathrm{PCl}_{5}(\mathrm{~g}) \rightleftharpoons \mathrm{PCl}_{3}(\mathrm{~g})+\mathrm{Cl}_{2}(\mathrm{~g})
$$

At equilibrium 0.05
$\mathrm{x} \quad \mathrm{x}$
$\mathrm{K}_{\mathrm{c}}=8.3 \times 10^{-3}=\frac{\left[\mathrm{PCl}_{3}\right]\left[\mathrm{Cl}_{2}\right]}{\left[\mathrm{PCl}_{5}\right]}$
$8.3 \times 10^{-3}=\frac{\mathrm{x}^{2}}{0.05}$
$x^{2}=0.415 \times 10^{-3}=4.15 \times 10^{-4}$
$\mathrm{x}=2.037 \times 10^{-2}=2.04 \times 10^{-2} \mathrm{~mol} \mathrm{~L}^{-1}$
Hence, $\left[\mathrm{PCl}_{3}\right]=\left[\mathrm{Cl}_{2}\right]=2.04 \times 10^{-2} \mathrm{~mol} \mathrm{~L}^{-1}$.

Number of stereoisomers possible for the following compound is


Solution

(1)
(2)

(3)

1,2 have plane of symmetry and are optically inactive. 3 is optically active as there is no plane of symmetry.
22. In how many of the following reactions, one of the products obtained is a yellow coloured precipitate?
(i) $\mathrm{Pb}^{2+}+\mathrm{KI} \longrightarrow$
(ii) $\mathrm{Pb}^{2+}+\mathrm{H}_{2} \mathrm{SO}_{4} \longrightarrow$
(iii) $\mathrm{Al}^{3+}+\mathrm{NaOH} \longrightarrow$
(iv) $\mathrm{Mg}^{2+}+\mathrm{Na}_{2} \mathrm{HPO}_{4} \longrightarrow$
(v) $\mathrm{Pb}^{2+}+\mathrm{K}_{2} \mathrm{CrO}_{4} \longrightarrow$
(vi) $\mathrm{I}^{-}+\mathrm{AgNO}_{3} \longrightarrow$
(vii) $\mathrm{SO}_{4}^{2-}+\mathrm{BaCl}_{2} \longrightarrow$

Solution

Reactions (i), (v) and (vi) give yellow ppt. of $\mathrm{PbI}_{2}, \mathrm{PbCrO}_{4}$ and AgI respectively.

How many $2^{\circ}$ carbon in the following?


Solution

There are 21 carbon atoms

24 The dipole moment of HBr is $1.6 \times 10^{-30}$ Coloumb-metre and inter-atomic spacing is $1 \stackrel{\circ}{\mathrm{~A}}$. The \% ionic character of HBr is

Solution

Charge of electron $=1.6 \times 10^{-19} \mathrm{C}$
Dipole moment of $\mathrm{HBr}=1.6 \times 10^{-30} \mathrm{C} \times \mathrm{m}$
Inter-atomic space (distance) $=1 \AA$
$=1 \times 10^{-10} \mathrm{~m}$
Percentage of ionic character in HBr
$=\frac{\text { Dipole moment of } \mathrm{HBr} \times 100}{\text { inter-atomic distance } \times q}$
$=\frac{1.6 \times 10^{-30}}{1.6 \times 10^{-19} \times 10^{-10}} \times 100$
$=10^{-30} \times 10^{29} \times 100$
$=10^{-1} \times 100$
$=0.1 \times 100$
$=10 \%$

25 How many complexes among the following are paramagnetic
$\left[\mathrm{Mn}(\mathrm{CN})_{6}\right]^{3-}\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+}\left[\mathrm{Co}(\mathrm{en})_{3}\right]^{3+}$
$\left[\mathrm{V}(\mathrm{CO})_{6}\right],\left[\mathrm{Ni}\left(\mathrm{NH}_{3}\right)_{6}\right]^{2+}\left[\mathrm{Ni}(\mathrm{dmg})_{2}\right]$
$\left[\mathrm{Pt}(\mathrm{Cl})_{2}\left(\mathrm{NH}_{3}\right)_{2}\right]\left[\mathrm{Cu}\left(\mathrm{NH}_{3}\right)_{4}\right]^{2+}\left[\mathrm{Cu}(\mathrm{CN})_{4}\right]^{3-}$

## Solution

$\left[\mathrm{Mn}(\mathrm{CN})_{6}\right]^{3-},\left[\mathrm{Cr}\left(\mathrm{H}_{2} \mathrm{O}\right)_{6}\right]^{3+},\left[\mathrm{V}(\mathrm{CO})_{6}\right],\left[\mathrm{Ni}\left(\mathrm{NH}_{3}\right)_{6}\right]^{2+},\left[\mathrm{Cu}\left(\mathrm{NH}_{3}\right)_{4}\right]^{2+}$
are paramagnetic due to the presence of unpaired electrons.
mathematics
1 If the cube roots of unity are $1, w, w^{2}$, then the roots of the equation $(x-1)^{3}+8=0$ are

Solution

Given equation is $(x-1)^{3}+8=0$
$\Rightarrow x=1-2(1)^{1 / 3}$
$\Rightarrow x=1-2,1-2 \omega, 1-2 \omega^{2} \quad$ (since $1, w, w^{2}$ are cube roots of unity)
$\Rightarrow x=-1,1-2 w, 1-2 w^{2}$
$2 \frac{d}{d x}\left\{\begin{array}{c}\tan ^{-1}\left(\frac{2 x}{1-x^{2}}\right)+\tan ^{-1}\left(\frac{3 x-x^{3}}{1-3 x^{2}}\right) \\ -\tan ^{-1}\left(\frac{4 x-4 x^{3}}{1-6 x^{2}+x^{4}}\right)\end{array}\right\}$ is equal to $\{|x|<\sqrt{2}-1\}$

## Solution

Let $I=\frac{d}{d x}\left\{\begin{array}{c}\tan ^{-1}\left(\frac{2 x}{1-x^{2}}\right)+\tan ^{-1}\left(\frac{3 x-x^{3}}{1-3 x^{2}}\right) \\ -\tan ^{-1}\left(\frac{4 x-4 x^{3}}{1-6 x^{2}+x^{4}}\right)\end{array}\right\}$

Put $x=\tan \theta$ the given equation
$\therefore \quad I=\frac{d}{d x}\left\{\tan ^{-1}(\tan 2 \theta)+\tan ^{-1}(\tan 3 \theta)-\tan ^{-1}(\tan 4 \theta)\right\}$
$=\frac{d}{d x}(\theta)=\frac{d}{d x}\left(\tan ^{-1} x\right)=\frac{1}{1+x^{2}}$

3 If $\tan \alpha=\left(1+2^{-x}\right)^{-1}, \tan \beta=\left(1+2^{x+1}\right)^{-1}$, then $(\alpha+\beta)$ can be equal to
Solution

We have, $\tan (\alpha+\beta)=\frac{\tan \alpha+\tan \beta}{1-\tan \alpha \tan \beta}$
$\because \tan \alpha=\frac{1}{1+2^{-x}}$ and $\tan \beta=\frac{1}{1+2^{x+1}}$
$\therefore \tan (\alpha+\beta)=\frac{\frac{1}{1+\frac{1}{2^{x}}}+\frac{1}{1+2^{x+1}}}{1-\frac{1}{1+\frac{1}{2^{x}}} \cdot \frac{1}{1+2^{x+1}}}$
$\Rightarrow \tan (\alpha+\beta)=\frac{2^{x}+2 \cdot 2^{2 x}+2^{x}+1}{1+2^{x}+2 \cdot 2^{x}+2 \cdot 2^{2 x}-2^{x}}$
$\Rightarrow \tan (\alpha+\beta)=1$
$\Rightarrow \alpha+\beta=\frac{\pi}{4}$

4 If $\tan ^{-1} x+\tan ^{-1} y=\frac{\pi}{4}$, then

## Solution

Given, $\tan ^{-1} x+\tan ^{-1} y=\frac{\pi}{4}$

Taking tan on both sides
$\left(\frac{x+y}{1-x y}\right)=1$
$\Rightarrow \quad \frac{x+y}{1-x y}=1$
$\Rightarrow \quad x+y+x y=1$Let two numbers have an arithmetic mean 9 and geometric mean 4 , then these numbers are the roots of the quadratic equation

Solution
$\therefore \frac{\alpha+\beta}{2}=9$ and $\sqrt{\alpha \beta}=4 \Rightarrow \alpha+\beta=18, \alpha \beta=16$
$\therefore$ Required equation is
$x^{2}-(\alpha+\beta) x+\alpha \beta=0$
$\Rightarrow x^{2}-18 x+16=0$
6. $\lim _{x \rightarrow 0} \frac{\cos 2 x-1}{\cos x-1}=$

Solution
$\lim _{x \rightarrow 0} \frac{\cos 2 x-1}{\cos x-1}=\lim _{x \rightarrow 0} \frac{\left(2 \cos ^{2} x-1\right)-1}{\cos x-1}$
$=\lim _{x \rightarrow 0} \frac{2\left(\cos ^{2} x-1\right)}{\cos x-1}$
$=\lim _{x \rightarrow 0} 2\left[\frac{(\cos x+1)(\cos x-1)}{(\cos x-1)}\right]$
$=2 \lim _{x \rightarrow 0}(\cos x+1)=2(1+1)=4$

7 Area bounded by the curves $y=\log _{e} x$ and $y=\left(\log _{e} x\right)^{2}$ is
Solution

Given curves are $\mathrm{y}=\log _{\mathrm{e}} \mathrm{x}$ and $\mathrm{y}=\left(\log _{e} \mathrm{x}\right)^{2}$
Solving $\log _{\mathrm{e}} \mathrm{x}=\left(\log _{\mathrm{e}} \mathrm{x}\right)^{2} \Rightarrow \log _{\mathrm{e}} \mathrm{x}=0,1 \Rightarrow \mathrm{x}=1$ and $\mathrm{x}=\mathrm{e}$
Also, for $1<\mathrm{x}<\mathrm{e}, 0<\log _{e} \mathrm{x}<1 \Rightarrow \log _{e} \mathrm{x}>\left(\log _{e} \mathrm{x}\right)^{2}$
For $\mathrm{x}>\mathrm{e}, \log _{\mathrm{e}} \mathrm{x}<\left(\log _{e} \mathrm{x}\right)^{2}$
$y=\left(\log _{e} x\right)^{2}>0$ for all $x>0$
And when $\mathrm{x} \rightarrow 0,\left(\log _{\mathrm{e}} \mathrm{x}\right)^{2} \rightarrow \infty$
From these information, we can plot the graph of the functions.


Then the required area $=\int_{1}^{e}\left(\log x-\left(\log _{e} x\right)^{2}\right) d x$
$=\int_{1}^{\mathrm{e}} \log \mathrm{xdx}-\int_{1}^{\mathrm{e}}\left(\log _{\mathrm{e}} \mathrm{x}\right)^{2} \mathrm{dx}$
$=\left[x \log _{e} x-x\right]_{1}^{e}-\left[x\left(\log _{e} x\right)^{2}\right]_{1}^{e}+\int_{1}^{e} \frac{2 \log _{e} x}{x} x d x$
$=1-\mathrm{e}+2\left[\operatorname{xlog}_{\mathrm{e}} \mathrm{x}-\mathrm{x}\right]_{1}^{\mathrm{e}}=3-\mathrm{e}$ sq. units

8 The solution of the differential equation $\frac{d y}{d x}+\frac{2 x}{1+x^{2}} \cdot y=\frac{1}{\left(1+x^{2}\right)^{2}}$ is
Solution

Given, $\frac{d y}{d x}+\frac{2 x}{1+x^{2}} \cdot y=\frac{1}{\left(1+x^{2}\right)^{2}}$
$\therefore$ IF $=e^{\int \frac{2 x}{1+x^{2}} d x}=e^{\log \left(1+x^{2}\right)}=1+x^{2}$

The complete solution is
$y\left(1+x^{2}\right)=\int\left(1+x^{2}\right) \cdot \frac{1}{\left(1+x^{2}\right)^{2}} d x+c$
$\Rightarrow y\left(1+x^{2}\right)=\tan ^{-1} x+c$

9 Let $A B$ be a given chord of the circle $x^{2}+y^{2}=r^{2}$ subtending a right angle at the centre. Then, the locus of the centroid of the $\Delta P A B$ as $P$ moves on the circle is

## Solution

Given equation of circle is $x^{2}+y^{2}=r^{2}$. Let any point on the circle is $P(r \cos \theta, r \sin \theta)$ and let the coordinates of centriod of the triangle be $(\alpha, \beta)$


Then, $\alpha=\frac{r+r \cos \theta}{3}$
$\Rightarrow \quad \frac{r}{3} \cos \theta=\alpha-\frac{r}{3}$
and $\beta=\frac{r+r \sin \theta}{3}$
$\Rightarrow \frac{r}{3} \sin \theta=\beta-\frac{r}{3}$
Now, $\left(\alpha-\frac{r}{3}\right)^{2}+\left(\beta-\frac{r}{3}\right)^{2}=\frac{r^{2}}{9}$
$\therefore$ The locus is $\left(x-\frac{r}{3}\right)^{2}+\left(y-\frac{r}{3}\right)^{2}=\left(\frac{r}{3}\right)^{2}$ which is a circle

If $\mathrm{A}, \mathrm{B}, \mathrm{C}$ are the angles of a triangle and $\left|\begin{array}{lll}1 & 1 & 1 \\ 1+\sin \mathrm{A} & 1+\sin \mathrm{B} & 1+\sin \mathrm{C} \\ \sin \mathrm{A}+\sin ^{2} \mathrm{~A} & \sin \mathrm{~B}+\sin ^{2} \mathrm{~B} & \sin \mathrm{C}+\sin ^{2} \mathrm{C}\end{array}\right|=0$, then which of the following is never true for triangle ABC ?
$\mathrm{R}_{2} \rightarrow \mathrm{R}_{2}-\mathrm{R}_{1}$
$\left|\begin{array}{lll}1 & 1 & 1 \\ \sin \mathrm{~A} & \sin \mathrm{~B} & \sin \mathrm{C} \\ \sin \mathrm{A}+\sin ^{2} \mathrm{~A} & \sin \mathrm{~B}+\sin ^{2} \mathrm{~B} & \sin \mathrm{C}+\sin ^{2} \mathrm{C}\end{array}\right|$
$\mathrm{R}_{3} \rightarrow \mathrm{R}_{3}-\mathrm{R}_{2}$
$=\left|\begin{array}{lll}1 & 1 & 1 \\ \sin \mathrm{~A} & \sin \mathrm{~B} & \sin \mathrm{C} \\ \sin ^{2} \mathrm{~A} & \sin ^{2} \mathrm{~B} & \sin ^{2} \mathrm{C}\end{array}\right|$
$=(\sin \mathrm{A}-\sin \mathrm{B}) \times(\sin \mathrm{B}-\sin \mathrm{C}) \times(\sin \mathrm{C}-\sin \mathrm{A})$
$\Rightarrow \mathrm{A}=\mathrm{B}$ or $\mathrm{B}=\mathrm{C}$ or $\mathrm{C}=\mathrm{A}$

So $\Delta$ is isosceles or equilateral

The acute angle of intersection between the curves $y=\sin x$ and $y=\cos x$ is
Solution

Equation of given curves are
$y=\sin x \ldots$ (i)
and $y=\cos x \ldots$..(ii)
On solving Eqs. (i) and (ii), we get
$x=\frac{\pi}{4}$
$\therefore$ Point of intersection of curves is $\left(\frac{\pi}{4}, \frac{1}{\sqrt{2}}\right)$
For $y=\sin x, \frac{d y}{d x}=\cos x$
(say)
For $y=\cos x \quad \Rightarrow \frac{d y}{d x}=-\sin x$
(say)
$\therefore \tan \theta=\frac{m_{1}-m_{2}}{1+m_{1} m_{2}}=\frac{\frac{1}{\sqrt{2}}+\frac{1}{\sqrt{2}}}{1-\frac{1}{\sqrt{2}} \cdot \frac{1}{\sqrt{2}}}=\frac{\frac{2}{\sqrt{2}}}{\frac{1}{2}}$
$\Rightarrow \tan \theta=2 \sqrt{2} \Rightarrow \theta=\tan ^{-1}(2 \sqrt{2})$

12 Let $\mathrm{x}, \mathrm{y}$ and z be the respective sum of the first n terms, the next n terms and the next n terms of a geometric progression, then $\mathrm{x}, \mathrm{y}, \mathrm{z}$ are in

## Solution

Let A be the $1^{\text {st }}$ term and R be the common ratio of the given G.P.
Sum of the first $n$ terms of the G.P. is $x=\frac{A\left(1-R^{n}\right)}{1-R}$. The next $n$ terms form a G.P. series of $n$ terms with the first term as the $(n+1)^{\text {th }}$ term of the given G.P. is given by $t_{n+1}=A R^{n}$. Sum of the next $n$ terms of the G.P. is $y \Rightarrow y=\frac{A R^{n}\left(1-R^{n}\right)}{1-R}$ and
$\mathrm{z}=\mathrm{AR}^{2 \mathrm{n}} \frac{\left(1-\mathrm{R}^{\mathrm{n}}\right)}{1-\mathrm{R}} \Rightarrow \mathrm{y}^{2}=\mathrm{zx}$
$\therefore \mathrm{x}, \mathrm{y}, \mathrm{z}$ are in G.P.
(13 If ${ }^{18} C_{15}+2\left({ }^{18} C_{16}\right)+{ }^{17} C_{16}+1={ }^{n} C_{3}$, then $n$ is equal to
Solution
${ }^{18} C_{15}+2\left({ }^{18} C_{16}\right)+{ }^{17} C_{16}+1={ }^{n} C_{3}$
$\Longrightarrow{ }^{18} C_{15}+{ }^{18} C_{16}+{ }^{18} C_{16}+{ }^{17} C_{16}+{ }^{17} C_{17}={ }^{n} C_{3}$
$\Longrightarrow \quad{ }^{19} C_{16}+{ }^{18} C_{16}+{ }^{18} C_{17}={ }^{n} C_{3}$
$\Longrightarrow \quad{ }^{19} C_{16}+{ }^{19} C_{17}={ }^{n} C_{3}$
$\Longrightarrow{ }^{20} C_{17}={ }^{n} C_{3} \Longrightarrow{ }^{20} C_{3}={ }^{n} C_{3} \Longrightarrow n=20$

Through a point $A$ on the $x$ - axis a straight line is drawn parallel to $y$ - axis so as to meet the pair of straight lines $\mathrm{ax}^{2}+2 h x y+$ $b^{2}=0$ in $B$ and $C$. If $A B=B C$ then :

Solution

$\mathrm{AB}=\mathrm{BC} \Rightarrow \mathrm{AC}=2 \mathrm{AB}$. Let $\mathrm{AB}=\mathrm{y}$,
$\Rightarrow \mathrm{by}^{2}+2 \mathrm{hx}_{1} \mathrm{y}+\mathrm{ax}_{1}^{2}=0$ has roots $\mathrm{y}_{1}, 2 \mathrm{y}_{1}$
$3 \mathrm{y}_{1}=-\frac{2 \mathrm{hx}_{1}}{\mathrm{~b}}, 2 \mathrm{y}_{1}^{2}=\frac{\mathrm{ax}_{1}^{2}}{\mathrm{~b}}$
$\Rightarrow \quad \frac{9 \mathrm{y}_{1}^{2}}{2 \mathrm{y}_{1}^{2}}=\frac{4 \mathrm{~h}^{2}}{\mathrm{~b}^{2}} \frac{\mathrm{x}_{1}^{2} \mathrm{~b}}{\mathrm{ax}} \frac{x_{1}^{2}}{}$
$\Rightarrow \quad \frac{9}{2}=\frac{4 \mathrm{~h}^{2}}{\mathrm{ab}} \Rightarrow 9 \mathrm{ab}=8 \mathrm{~h}^{2}$

The greatest integer which divides the number $101^{100}-1$ is

Solution
$(1+100)^{100}=1+100.100+\frac{100.99}{1.2}(100)^{2}+\ldots \Rightarrow 101^{100}-1=100.100\left[1+\frac{100.99}{1.2}+\frac{100.99 .98}{3 \cdot 2.1} 100+\ldots.\right]$
From above it is clear that, $101^{100}-1$ is divisible by $(100)^{2}=10000$

In an steamer there are stalls for 12 animals and there are horses, cows and calves (not less then 12 each) ready to be shipped. The number of ways, the ship load can be made is ...

Solution

First stall can be filled in 3 ways, second stall can be filled in 3 ways and so on.

$$
=3 \times 3 \times \ldots \times 3(12 \text { times })=3^{12}
$$

17
The domain of the function
$f(x)=\frac{\sin ^{-1}(x-3)}{\sqrt{9-x^{2}}}$ is

## Solution

The function $f(x)$ will be defined, if

$$
-1 \leq(x-3) \leq 1 \Rightarrow 2 \leq x \leq 4
$$

And $9-x^{2}>0 \Rightarrow-3<x<3$
$\therefore \quad 2 \leq x<3$

Let $I=\int_{0}^{1} \frac{\sin x}{\sqrt{x}} d x$ and $J=\int_{0}^{1} \frac{\cos x}{\sqrt{x}} d x$. Then, which one of the following is true?

Solution

Since, $I=\int_{0}^{1} \frac{\sin x}{\sqrt{x}} d x<\int_{0}^{1} \frac{x}{\sqrt{x}} d x$
as in $x \in(0,1), x>\sin x$
$I<\int_{0}^{1} \sqrt{x} d x=\frac{2}{3}\left[x^{3 / 2}\right]_{0}^{1} \Rightarrow I<\frac{2}{3}$
For, $x \in(0,1), \frac{\cos x}{\sqrt{x}}<\frac{1}{\sqrt{x}}$
Hence, $J=\int_{0}^{1} \frac{\cos x}{\sqrt{x}} d x<\int_{0}^{1} x^{-\frac{1}{2}} d x=2 \Rightarrow J<2$

19
From a 60 meter high tower angles of depression of the top and bottom of a house are $\alpha$ and $\beta$ respectively. If the height of the house is $\frac{60 \sin (\beta-\alpha)}{x}$, then $x=$

Solution
$H=d \tan \beta$ and $H-h=d \tan \alpha$
$\Rightarrow \frac{60}{60-h}=\frac{\tan \beta}{\tan \alpha} \Rightarrow-h=\frac{60 \tan \alpha-60 \tan \beta}{\tan \beta}$

$\Rightarrow h=\frac{60 \sin (\beta-\alpha)}{\cos \alpha \cos \beta \frac{\sin \beta}{\cos \beta}} \Rightarrow x=\cos \alpha \sin \beta$.
20. Out of $3 n$ consecutive natural numbers, 3 natural numbers are chosen at random without replacement. The probability that the sum of the chosen numbers is divisible by 3 , is

Solution

In $3 n$ consecutive natural numbers, either
(i) $n$ numbers are of form $3 P$
(ii) $n$ numbers are of form $3 P+1$
(iii) $n$ numbers are of form $3 P+2$

Here favourable number of cases= Either we can select three numbers from any of the set or we can select one from each set

Hence favourable number of cases
$={ }^{n} C_{3}+{ }^{n} C_{3}+{ }^{n} C_{3}+\left({ }^{n} C_{1} \times{ }^{n} C_{1} \times{ }^{n} C_{1}\right)$
$=3\left(\frac{n(n-1)(n-2)}{6}\right)+n^{3}$
$=\frac{n(n-1(n-2)}{2}+n^{3}$

Total number of selections $={ }^{3 n} C_{3}$
$\therefore$ Required probability
$=\frac{\frac{n(n-1)(n-2)}{2}+n^{3}}{\frac{3 n(3 n-1)(3 n-2)}{6}}$
$=\frac{3 n^{2}-3 n+2}{(3 n-1)(3 n-2)}$

Method - II

Let $3 n$ consecutive natural numbers are
$1,2,3, \ldots .3 n$
then let $s_{1}=\{1,4,7,10$ $\qquad$ $3 n-2\}$
$s_{2}=\{2,5,8,11$, ...... $3 n-1\}$
$s_{3}=\{3,6,9$ $\qquad$ $3 n\}$

Hence for sum of 3 numbers divisible by 3 , we can either select 3 numbers from $s_{1}$ or $s_{2}$ or $s_{3}$ or 1 number from each set $=n_{c_{3}}+n_{c_{3}}+n_{c_{3}}+n_{c_{i}} n_{c_{i}} n_{c_{i}}$
$=\frac{3 \cdot n(n-1)(n-2)}{6}+n^{3}$
$\frac{n\left(n^{2}-3 n+2+2 n^{2}\right)}{2}$
$\frac{n\left(3 n^{2}-3 n+2\right)}{2}$
and we can select 3 numbers by $3 n_{c_{3}}$ ways.
Hence required probability $=\frac{n\left(3 n^{2}-3 n+2\right)}{2}$
$\frac{3 n(3 n-1)(3 n-2)}{6}$
$={ }^{n} C_{3}+{ }^{n} C_{3}+{ }^{n} C_{3}+{ }^{n} C_{1} \cdot{ }^{n} C_{1} \cdot{ }^{n} C_{1}$

The area (in sq. units) bounded by the curve $y=\max .\left(x^{3}, x^{4}\right)$ and the $x$-axis from $x=0$ to $x=1$ is

Solution
$\max \left(x^{3}, x^{4}\right)=x^{3}(\forall x \in(0,1))$
$\therefore$ Required area $=\int_{0}^{1} x^{3} d x=\left(\frac{x^{4}}{4}\right)_{0}^{1}=\frac{1}{4}=0.25$ sq. units

If the value of $\left(1+\tan 1^{\circ}\right)\left(1+\tan 2^{\circ}\right)\left(1+\tan 3^{\circ}\right) \ldots \ldots \ldots\left(1+\tan 44^{\circ}\right)\left(1+\tan 45^{\circ}\right)$ is $2^{\lambda}, \lambda \in N$ then sum of digits of number $\lambda$ is

Solution

If $A+B=45^{\circ}$, then
$\tan (A+B)=1 \because \tan (A+B)=\frac{\tan A+\tan B}{1-\tan A \tan B}$
$\Rightarrow \tan A+\tan B=1-\tan A \tan B$
$\Rightarrow(1+\tan A)(1+\tan B)=2$

LHS
$=\left[\left(1+\tan 1^{\circ}\right) \cdot\left(1+\tan 44^{\circ}\right)\right] \cdot\left[\left(1+\tan 2^{\circ}\right) \cdot\left(1+\tan 43^{\circ}\right)\right] \ldots\left[\left(1+\tan 45^{\circ}\right)\right]\left[\right.$ for each $\left.(1+\tan \theta)\left[1+\tan \left(\frac{\pi}{4}-\theta\right)=2\right]\right]$
$=2^{22} .(1+1)$
$=2^{23}$
$=2^{\lambda}$
then $\lambda=23$ hence sum of digits of $\lambda$ is $2+3$
$=5$

Let $f(x)=\frac{9 x}{25}+c, c>0$. If the curve $y=f^{-1}(x)$ passes through $\left(\frac{1}{4},-\frac{5}{9}\right)$ and $g(x)$ is the antiderivative of $f^{-1}(x)$ such that $g(0)=\frac{5}{2}$, then the value of $[g(1)]$ is, (where [.] represents the greatest integer function)

Solution
$y=f(x)=\frac{9 x}{25}+c_{1}$
$\Rightarrow x=\frac{25}{9}\left(y-c_{1}\right)=f^{-1}(y)$
$\Rightarrow f^{-1}(x)=\frac{25}{9}\left(x-c_{1}\right)$
The curve $y=f^{-1}(x)$ passes through $\left(\frac{1}{4},-\frac{5}{9}\right)$
$\Rightarrow-\frac{5}{9}=\frac{25}{9}\left(\frac{1}{4}-c_{1}\right)$
$\Rightarrow c_{1}=\frac{1}{4}+\frac{1}{5}=\frac{9}{20}$
$\Rightarrow f^{-1}(x)=\frac{25}{9}\left(x-\frac{9}{20}\right)=\frac{25 x}{9}-\frac{5}{4}$
$g(x)=\int f^{-1}(x) d x=\int\left(\frac{25 x}{9}-\frac{5}{4}\right) d x$
$=\frac{25}{9}\left(\frac{x^{2}}{2}\right)-\frac{5 x}{4}+c_{2}$
$g(0)=\frac{5}{2} \Rightarrow c_{2}=\frac{5}{2}$
$\Rightarrow g(x)=\frac{25}{9}\left(\frac{x^{2}}{2}\right)-\frac{5 x}{4}+\frac{5}{2}$
$\Rightarrow g(1)=\frac{25}{18}-\frac{5}{4}+\frac{5}{2}=\frac{50-45+90}{36}=\frac{95}{36}$
$\Rightarrow[g(1)]=2$

In the figure $P Q, P O_{1}$ and $O_{1} Q$ are the diameters of semicircles $C_{1}, C_{2}$ and $C_{3}$ with centres at $O_{1}, O_{2}$ and $O_{3}$ respectively and the circle $C_{4}$ touches the semicircles $C_{1}, C_{2}$ and $C_{3}$. If $P Q=24$ units and the area of the circle $C_{4}$ is $A$ sq. units, then the value of $\frac{8 \pi}{A}$ is equal to (here, $P O_{1}=O_{1} Q$ )


Solution


Let the point of contact of $C_{4} \& C_{1}$ is $A$, center of $C_{4}$ is $O_{4} \&$ radius is equal to $r$
$\Rightarrow A O_{1}=12 \Rightarrow O_{1} O_{4}=12-r$
Also, $O_{4} O_{3}=r+6$ and $O_{1} O_{3}=6$
$\Rightarrow(r+6)^{2}=(12-r)^{2}+36$
$\Rightarrow 36 r=144 \Rightarrow r=4 \Rightarrow A=16 \pi$
$\Rightarrow \frac{8 \pi}{A}=\frac{8 \pi}{16 \pi}=\frac{1}{2}=0.5$

Let f , g and h are differentiable functions. If $\mathrm{f}(0)=1 ; \mathrm{g}(0)=2 ; \mathrm{h}(0)=3$ and the derivatives of their pair wise products at $\mathrm{x}=0$ are $(\mathrm{fg})^{\prime}(0)=6 ;(\mathrm{gh})^{\prime}(0)=4$ and $(\mathrm{hf})^{\prime}(0)=5$, then the value of $(\mathrm{fgh})^{\prime}(0)$ is

Solution

Let, $\mathrm{y}=\mathrm{fgh}$

$$
\frac{\mathrm{dy}}{\mathrm{dx}}=\mathrm{f}^{\prime} \mathrm{gh}+\mathrm{fg}^{\prime} \mathrm{h}+\mathrm{fgh}^{\prime}
$$

$$
=\frac{1}{2}\left(2 \mathrm{f}^{\prime} \mathrm{gh}+2 \mathrm{fg}^{\prime} \mathrm{h}+2 \mathrm{fgh}^{\prime}\right)
$$

$$
=\frac{1}{2}\left(\mathrm{~h}\left(\mathrm{f}^{\prime} \mathrm{g}+\mathrm{g}^{\prime} \mathrm{f}\right)+\mathrm{g}\left(\mathrm{f}^{\prime} \mathrm{h}+\mathrm{fh}^{\prime}\right)+\mathrm{f}\left(\mathrm{~g}^{\prime} \mathrm{h}+\mathrm{gh}^{\prime}\right)\right)
$$

$$
=\frac{1}{2}\left[\mathrm{~h} .(\mathrm{fg})^{\prime}+\mathrm{g} \cdot(\mathrm{fh})^{\prime}+\mathrm{f} .(\mathrm{gh})^{\prime}\right]
$$

$(\mathrm{fgh})^{\prime}(0)=\frac{1}{2}\left[\mathrm{~h}(0)(\mathrm{fg})^{\prime}(0)+\mathrm{g}(0)(\mathrm{fh})^{\prime}(0)+\mathrm{f}(0)(\mathrm{gh})^{\prime}(0)\right]$

$$
=\frac{1}{2}(3 \times 6+2 \times 5+1 \times 4)
$$

$=\frac{1}{2}(18+10+4)=\frac{32}{2}=16$


[^0]:    11
    ${ }_{90} \mathrm{Th}^{228}$ emits four alpha and one beta particle. Number of neutrons in daughter element is

